

Letter

The GdPt₂Sn-type crystal structure of CePd₂In

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1. Introduction

Ternary compounds with the general composition RPt₂In and RPt₂Sn (R, rare earth) were investigated by Mooij and Buschow [1] and Dwight [2]. The crystal structure of these compounds was found to be of the hexagonal ZrPt₂Al type (space group *P6₃/mmc*), more correctly named the GdPt₂Sn type according to Parthé [3]. As part of an investigation into the crystal structures of cerium compounds of the form CeT₂X (T, transition element; X, B element), the crystal structure of CePd₂In has been determined by the Rietveld powder method.

2. Experimental details

The sample was synthesized by arc melting the constituent elements under purified argon in a water-cooled copper hearth. All elements had a nominal purity of 99.9%. Traces of foreign phases were detectable on the Guinier powder photograph taken with copper K α_1 radiation. Attempts to improve the phase purity by annealing at 580 and 600 °C for 2–4 days had little success. Indexing was verified with the aid of the Lazy-Pulverix programme [4]. Diffraction data were collected on a PADX diffractometer with Cu K α radiation in steps of 0.02° within the range 18° < 2 θ < 85°. For the structure refinement we used the X-ray Rietveld system XRS-82 [5]. A total of 22 structural parameters and nine profile parameters were varied during the course of the refinements. After deleting the foreign phases, the refinement finally converged to $R_F=0.047$, $R_{wp}=0.169$ and $R_{exp}=0.047$.

The magnetic susceptibility was measured on the polycrystalline sample between 2 and 300 K in a field of 10 kOe.

3. Results and discussion

The crystallographic data for CePd₂In and the relevant interatomic distances are given in Tables 1 and 2 respectively. The crystal structure is shown in Fig. 1. The weakest and strongest bonds are found between Ce and Pd and between In and Pd atoms respectively. The shortest bond length is In–Pd = 2.764 Å, equivalent to an 8.8% contraction with respect to the element radii [11]. The present bond lengths and angles are comparable with those of YPt₂In, where the shortest bond length is 2.746 Å between Pt and In, corresponding to the largest contraction of 10% [2]. As expected, the CePd₂In unit cell is slightly larger than that of YPt₂In. Similar Ce–Ce, In–In, Ce–In, Pd–Pd and Pd–In bond lengths are also found in related compounds such as Ce₂In [6] and LaPd₂In [7].

The magnetic measurements showed that the Curie–Weiss law is obeyed from about 100 to 300 K

TABLE 1. Crystallographic data for CePd₂In ($T=295$ K): GdPt₂Sn type, *P6₃/mmc* (No. 194), $Z=2$, $a=4.6272(5)$ Å, $c=9.1975(5)$ Å, $V=170.54(9)$ Å³, $d_x=9.107$ g cm⁻³

Element	Site	x	y	z	U_{eq}
Pd	4f	$\frac{1}{3}$	$\frac{2}{3}$	0.5775(1)	0.0166(4)
Ce	2c	$\frac{1}{3}$	$\frac{2}{3}$	$\frac{1}{2}$	0.0171(4)
In	2a	0	0	0	0.0160(0)

$$R_f=0.047, R_{wp}=0.169, R_{exp}=0.047.$$

TABLE 2. Interatomic distances (Å) for CePd₂In

Ce–2Pd	3.010(1)	Pd–3In	2.764(1)
6Pd	3.108(1)	1Ce	3.010(1)
6In	3.525(1)	3Pd	3.026(1)
In–6Pd	2.764(1)	3Ce	3.108(1)
6Ce	3.525(1)	1Pd	3.177(1)

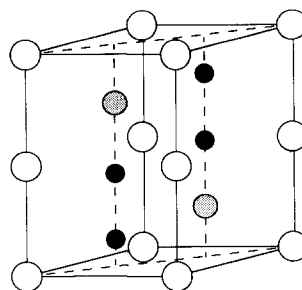


Fig. 1. Schematic representation of the hexagonal unit cell of GdPt₂Sn-type CePd₂In: ⊗, Ce in 2c; ○, In in 2a; ●, Pd in 4f.

with an effective Bohr magneton number $n_p = 2.57 \mu_B$ and a paramagnetic Curie temperature $\Theta_p = -2$ K. The experimental n_p fits well with the theoretical value of $2.54 \mu_B$. Small deviations from the Curie-Weiss law were observed below 100 K, demonstrating a weak crystal electric field influence on the $^2F_{5/2}$ ground state of the single 4f electron of cerium. No magnetic ordering was observed down to 2 K.

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